

# **Indoor Air Contaminant Monitoring & Removal**

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# **This Project Consists of 3 Related Tasks**

**All three tasks are concerned with the behavior of organic contaminants in indoor environments:**

- 1. Rapid Laboratory-Scale Testing of Contaminant Removal Performance of Desiccant Materials**
- 2. Contaminant Removal Performance Testing of Full-Scale TAT components**
- 3. Chemical Sensors for Real-time Indoor Air Quality Monitoring**

# Indoor Air Contaminant Monitoring & Removal

- 1. Rapid Laboratory-Scale Testing of Contaminant Removal Performance of Desiccant Materials**
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# Rapid Laboratory-Scale Testing of Contaminant Removal Performance of Desiccant Materials

Conventional Test methods were designed for Industrial Gas Cleaning Applications (e.g. carbon adsorption, zeolite concentration)

Industrial contaminants are typically present at 100+ part-per-million (ppm), while Indoor Environments typically have parts-per-billion (ppb) levels of organic contaminants

Application of conventional tests to ppb-level contaminants =excessively long tests (100's hours)

We have developed laboratory techniques to perform these tests up to 50 times faster than conventional techniques

Manuscript in preparation summarizing this work

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- 2. Contaminant Removal Performance Testing of Full-Scale TAT components**
3. Chemical Sensors for Real-time Indoor Air Quality Monitoring

# Contaminant Removal Performance Testing of Full-Scale TAT components

- New Task This Year
- Complements existing NREL expertise in thermal performance monitoring of TAT components
- Leveraging separate NREL Investments in state-of-the art GC/MS (Agilent 6890/5973) and Thermal Desorption Equipment (Perkin-Elmer ATD-400)
- Testing should commence this summer

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1. Rapid Laboratory-Scale Testing of Contaminant Removal Performance of Desiccant Materials
2. Contaminant Removal Performance Testing of Full-Scale TAT components
- 3. Chemical Sensors for Real-time Indoor Air Quality Monitoring**

# Background and Motivation

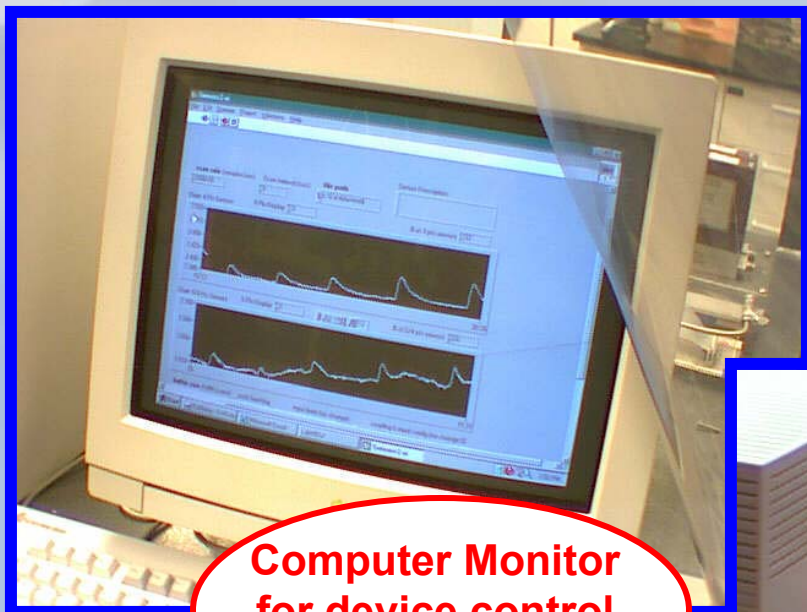
- Real-time monitoring is important for demand-controlled ventilation
- This is difficult because ambient concentrations are low (generally less than ppm)
- Analytical devices with these sensitivities are expensive (e.g. >\$25k- GC/MS, FTIR,...)
- Operating cost is high (know-how and maintenance)

## Our approach:

- Use low cost commercially-available sensors
- Exploit sophisticated signal processing to enhance sensor selectivity and sensitivity
- Package a “smart” unit



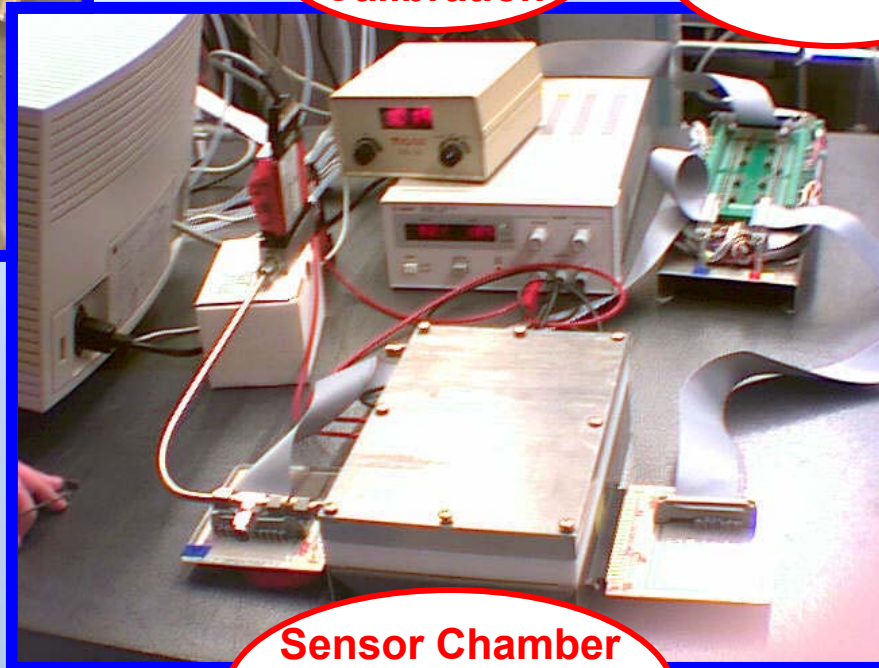
# Laboratory Development System



**Computer Monitor  
for device control  
and data output**



**Mass Flow  
Controllers for  
Calibration**

**Data Interface  
Card**



**Sensor Chamber  
RH, Temp,  
Hydrocarbon**

# FY01 Accomplishments:

- ✓ Design and construct prototype sensor system
- ✓ Design and construct data acquisition system
- ✓ Test overall system
- ✓ Develop a test protocol and data analysis strategy
- ✓ Characterize sensor geometry features
-  Characterize sensor signals
-  Demonstrate **qualitative** distinction among organic compounds
- ✓ Refine system geometry, modify sensor operating conditions

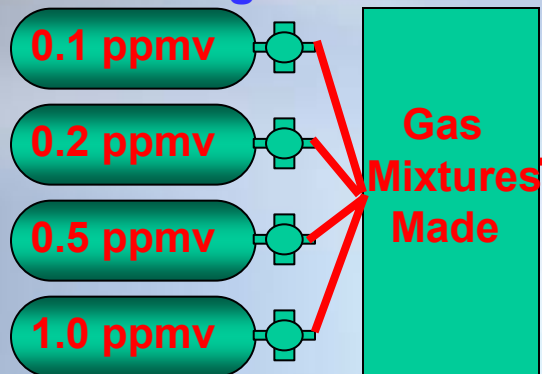
*Commercial sensors  
are good enough!  
Signal processing  
is powerful!*

# Tasks Completed **Since** Last June:

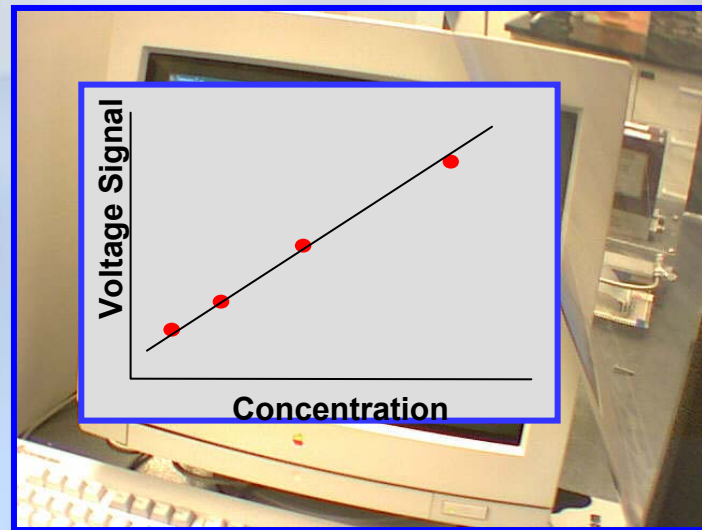
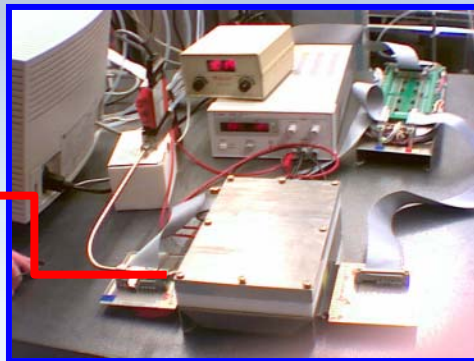
- ✓ Examined the effect of temperature
- ✓ Added temperature sensors to the system
- ✓ Added relative humidity sensor to the system
- ✓ Constructed a gas manifold for binary and ternary mixture calibrations
- ✓ Demonstrated the **generality** of the system calibration:  
toluene, benzene, isopropyl alcohol, acetone,  
ethanol, ethylbenzene, o-xylene
- ✓ Demonstrated binary and ternary **mixture quantitative** calibration
- ✓ Constructed a portable unit for demonstration and field tests

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Calibration  
gases

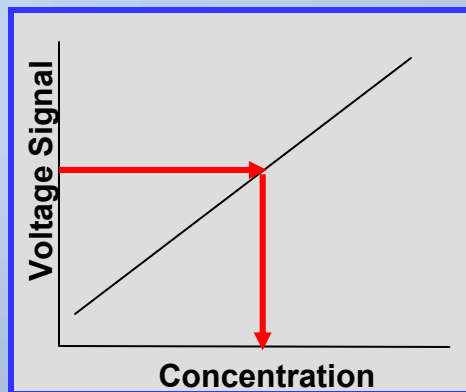
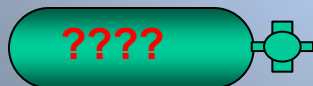


## Calibration



$$Y_{(\text{signal } V)} = b_0 + m_{\text{tol1}} X_{\text{tol}}$$

## Measurement (Inverse calibration)

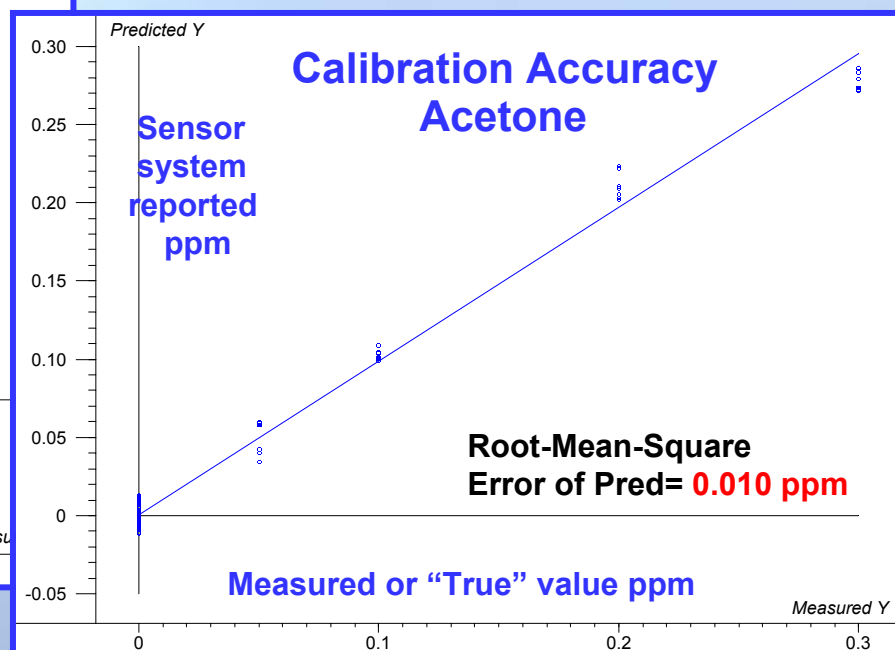
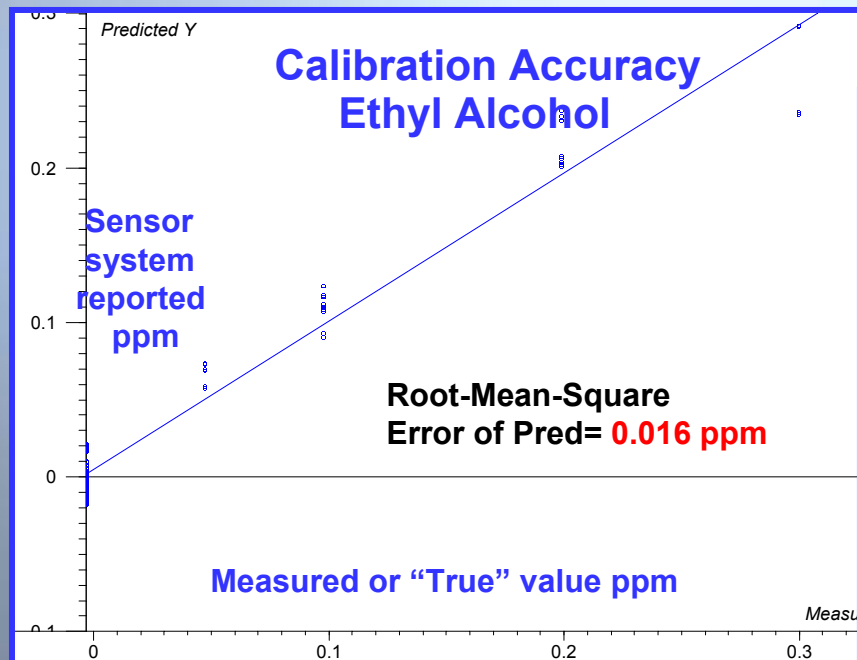
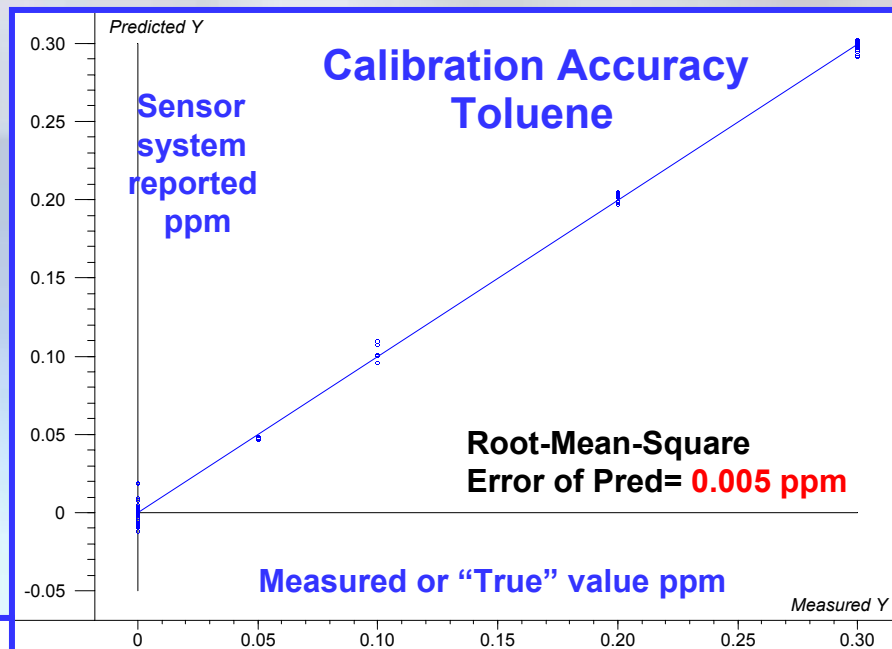


$$X_{\text{tol}} = \frac{Y_{\text{signal } V} - b_0}{m_{\text{tol1}}}$$

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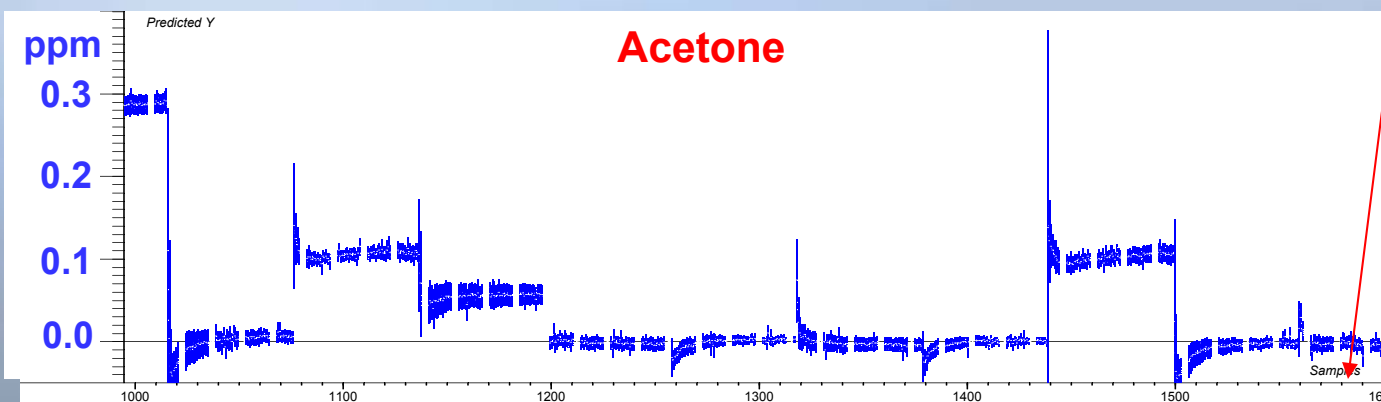
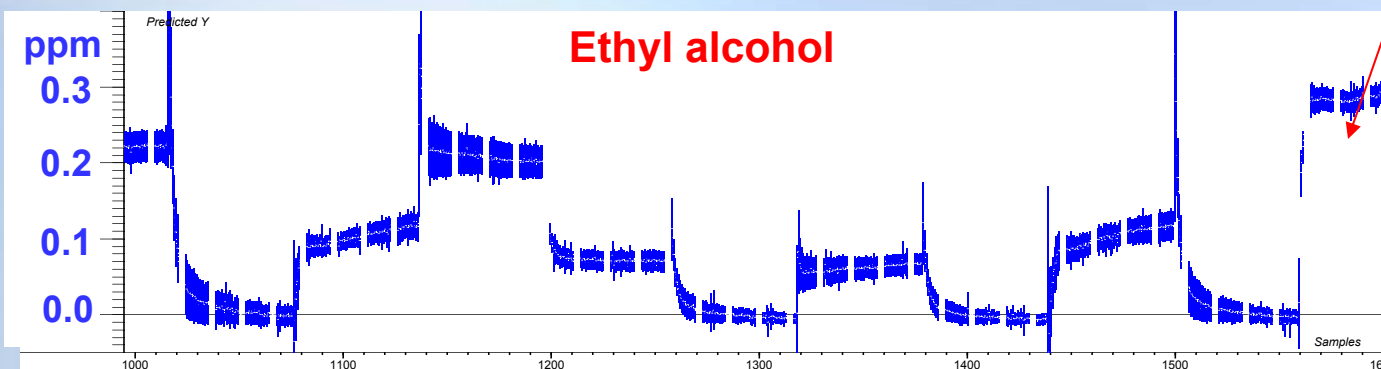
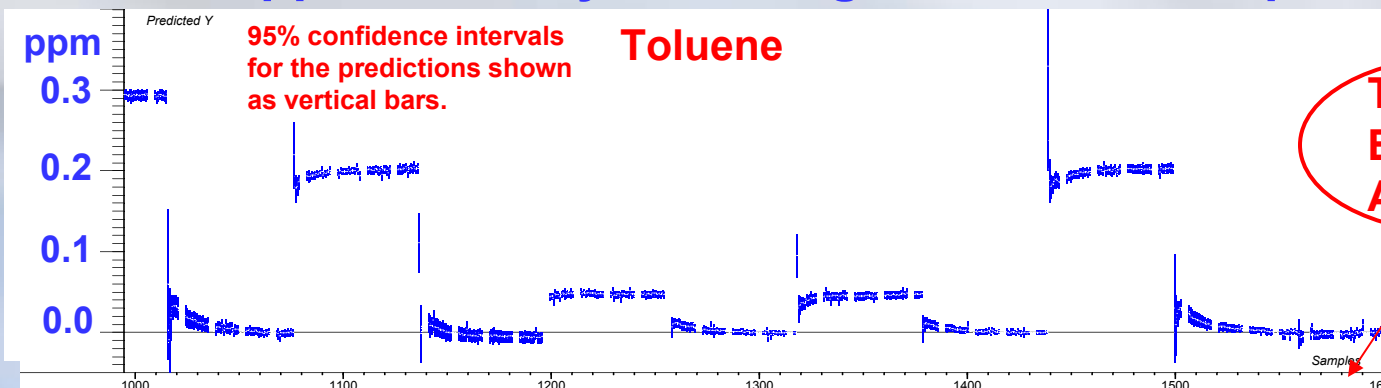
## Calibration accuracy in ternary mixtures

toluene, benzene, isopropyl alcohol,  
acetone, ethanol, ethylbenzene,  
o-xylene



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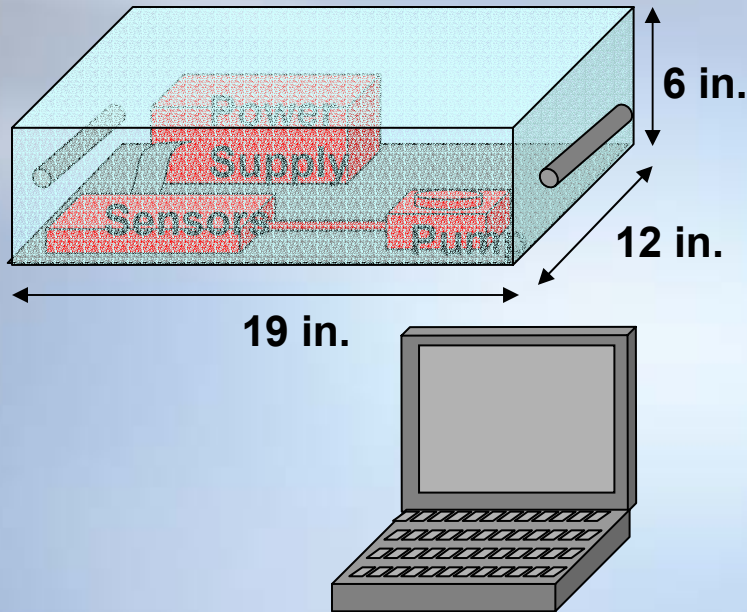
# Prediction of “apparent analyte” during a calibration experiment



Time (one measurement per 15 sec. shown)

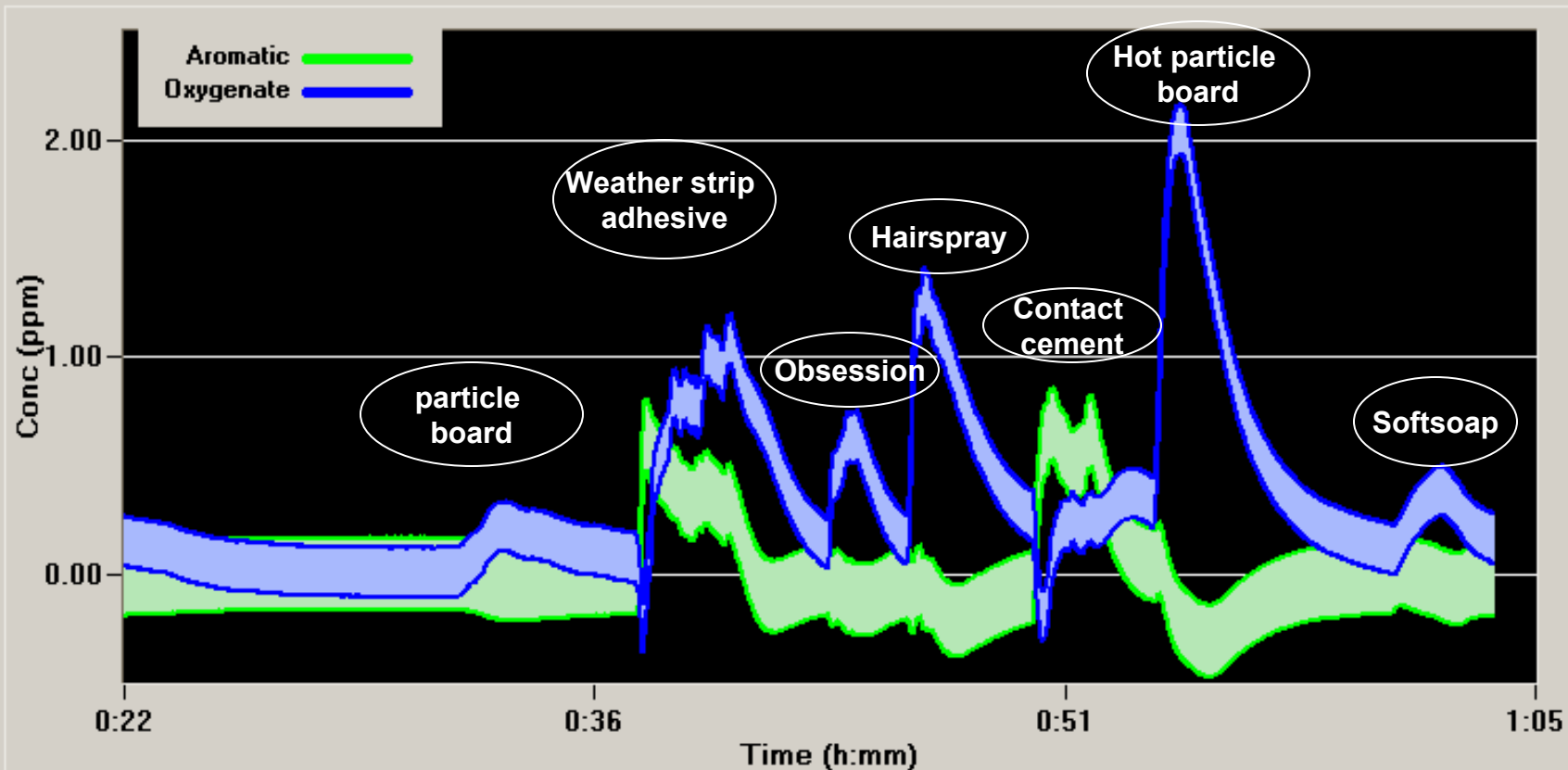


# 3 Portable Unit for Field Testing and Demonstration



## Test objectives:

- 1: to develop a self-contained control system (laptop)
- 2: to show that the system operates in different locations
- 3: to test whether calibration transfer (lab to unit) can be accomplished



Initialize

Start DAQ

End DAQ

Raw Log File: d:\RawLog.txt

Conc Log File: d:\ConcLog.txt

Calibration File: d:\Matrix-TEA2\_3pc.txt

Append

Overwrite

Time: 4/26/2002 12:12:53 PM

Interval (s) 1.0

X-Scale max: 65

Y-Scale max: 2.500

X-Scale min: 22

Y-Scale min: -0.500

VOC

End Demo

ReZero



# Vision of a Commercial Unit

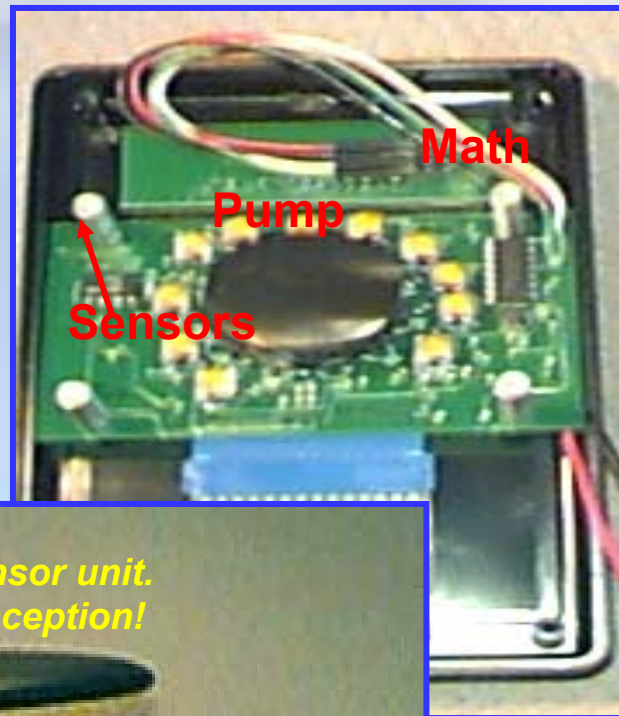
**What will it contain?**

**How will it function?**

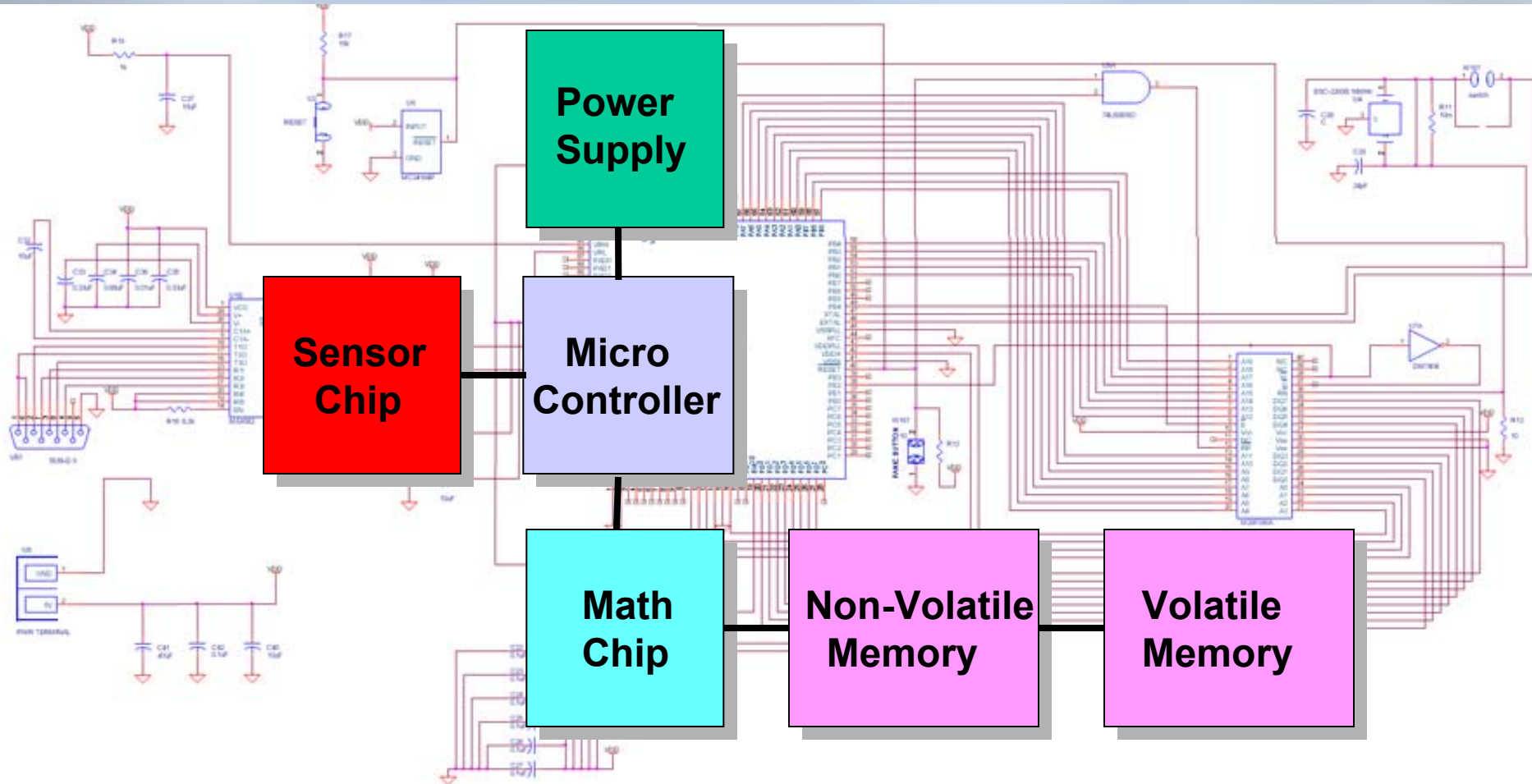
**Who might use this?**

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## Commercial unit: small, “smart”, self-contained unit



*Not an actual sensor unit.  
This is just a conception!*



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# Sensor and Memory Components

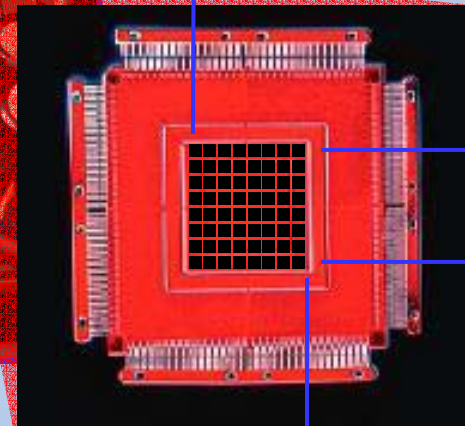
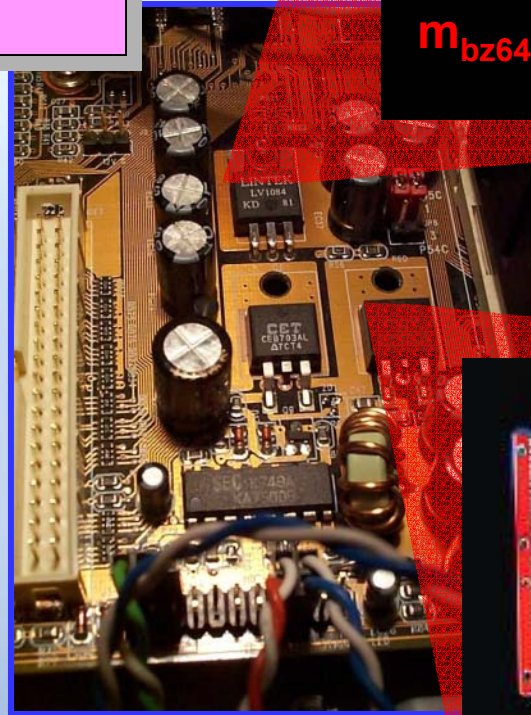
Calibration  
Coefficient  
Matrix

**Non-Volatile  
Memory**

Sensor signals  
over time

**Volatile  
Memory**

$m_{bz1}$	$m_{tol1}$	$m_{ac1}$	$m_{Z1}$
$m_{bz2}$	$m_{tol2}$	$m_{ac2}$	$m_{Z2}$
$m_{bz3}$	$m_{tol3}$	$m_{ac3}$	$\dots m_{Z3}$
$\vdots$	$\vdots$		
$m_{bz64}$	$m_{tol64}$	$m_{ac64}$	$m_{Z64}$

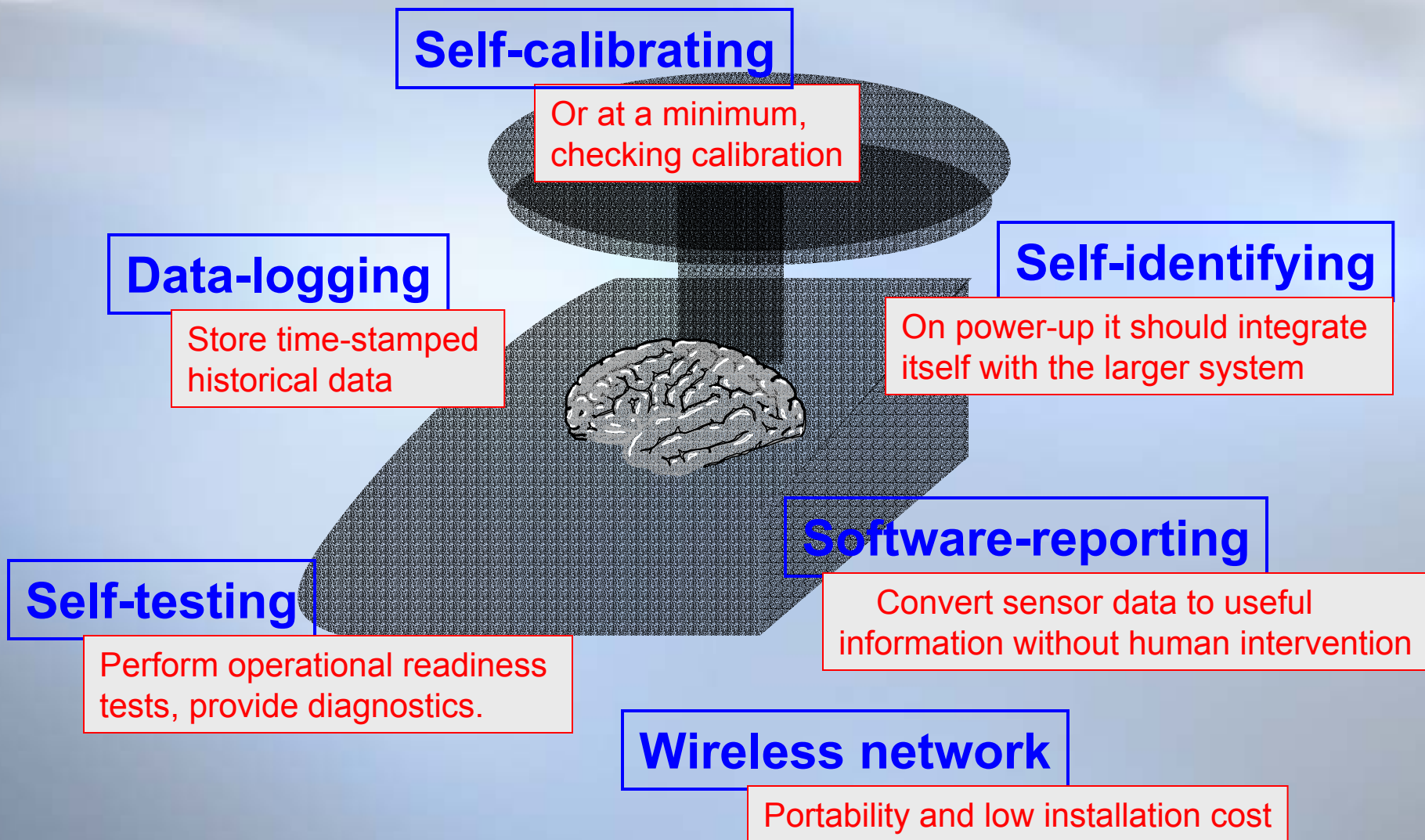


$V_1$



$V_{64}$



# Features of a “smart” commercial unit



# Next Steps

- Examine transient vs. steady state responses (validity of calibrations)
- Calibration transfer... A yellow starburst icon with a red outline and the text "Technology Hurdle" inside.
- Secure intellectual property
- Miniaturization of the solid state sensor A yellow starburst icon with a red outline and the text "Technology Hurdle" inside.